

## Average Atomic Mass Worksheet

**Standard 3.4** Compare characteristics of isotopes of the same element. Calculate the weighted average atomic mass of an element from isotopic abundance, given the atomic mass of each contributor

1. The element copper has naturally occurring isotopes with mass numbers of 63 and 65. The relative abundance and atomic masses are 69.2% for a mass of 62 amu and 30.8% for a mass of 64 amu. Calculate the average atomic mass of copper.
2. Calculate the average atomic mass of sulfur if 95.00% of all sulfur atoms have a mass of 31.972 amu, 0.76% has a mass of 32.971 amu and 4.22% have a mass of 33.967 amu.
3. The four isotopes of lead are shown below, each with its percent by mass abundance and the composition of its nucleus. Using the following data, first calculate the approximate atomic mass of each isotope. Then calculate the average atomic mass of lead.

82p	82p	82p	82p
122n	124n	125n	126n
1.37%	26.26%	20.82%	51.55%

4. Calculate the average atomic mass of bromine. One isotope of bromine has an atomic mass of 78.92 amu and a relative abundance of 50.69%. The other major isotope of bromine has an atomic mass of 80.92 amu and a relative abundance of 49.31%.
5. There are three isotopes of silicon. They have mass numbers of 28, 29 and 30. The average atomic mass of silicon is 28.086 amu. What does this say about the relative abundances of the three isotopes?

# Calculating Average Atomic Mass Worksheet Answers

**Robert Andrews Millikan**



Calculating Average Atomic Mass Worksheet Answers

*A New Empirical Mass Equation* Harris B. Levy, 1956

**Atomic and Molecular Weight Determination** Richard Bell Moyes, 1971

**Chemistry** Steven S. Zumdahl, Susan A. Zumdahl, 2012 Steve and Susan Zumdahl's texts focus on helping students build critical thinking skills through the process of becoming independent problem solvers. They help students learn to think like a chemist so they can apply the problem solving process to all aspects of their lives. In *CHEMISTRY AN ATOMS FIRST APPROACH 1e International Edition* the Zumdahls use a meaningful approach that begins with the atom and proceeds through the concept of molecules structure and bonding to more complex materials and their properties. Because this approach differs from what most students have experienced in high school courses it encourages them to focus on conceptual learning early in the course rather than relying on memorization and a plug and chug method of problem solving that even the best students can fall back on when confronted with familiar material. The atoms first organization provides an opportunity for students to use the tools of critical thinkers to ask questions to apply rules and models and to

Chemistry 2e Paul Flowers, Richard Langely, William R. Robinson, Klaus Hellmut Theopold, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course. The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them. The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning. The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition. Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative. Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition.

**University Physics** Samuel J. Ling, Jeff Sanny, William Moebs, 2017-12-19 University Physics is designed for the two or three semester calculus based physics course The text has been developed to meet the scope and sequence of most university physics courses and provides a foundation for a career in mathematics science or engineering The book provides an important opportunity for students to learn the core concepts of physics and understand how those concepts apply to their lives and to the world around them Due to the comprehensive nature of the material we are offering the book in three volumes for flexibility and efficiency Coverage and Scope Our University Physics textbook adheres to the scope and sequence of most two and three semester physics courses nationwide We have worked to make physics interesting and accessible to students while maintaining the mathematical rigor inherent in the subject With this objective in mind the content of this textbook has been developed and arranged to provide a logical progression from fundamental to more advanced concepts building upon what students have already learned and emphasizing connections between topics and between theory and applications The goal of each section is to enable students not just to recognize concepts but to work with them in ways that will be useful in later courses and future careers The organization and pedagogical features were developed and vetted with feedback from science educators dedicated to the project VOLUME II Unit 1 Thermodynamics Chapter 1 Temperature and Heat Chapter 2 The Kinetic Theory of Gases Chapter 3 The First Law of Thermodynamics Chapter 4 The Second Law of Thermodynamics Unit 2 Electricity and Magnetism Chapter 5 Electric Charges and Fields Chapter 6 Gauss's Law Chapter 7 Electric Potential Chapter 8 Capacitance Chapter 9 Current and Resistance Chapter 10 Direct Current Circuits Chapter 11 Magnetic Forces and Fields Chapter 12 Sources of Magnetic Fields Chapter 13 Electromagnetic Induction Chapter 14 Inductance Chapter 15 Alternating Current Circuits Chapter 16 Electromagnetic Waves

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**POGIL Activities for High School Chemistry** High School POGIL Initiative, 2012

*Fundamentals of General, Organic, and Biological Chemistry* John McMurry, 2013 *Fundamentals of General Organic and Biological Chemistry* by McMurry Ballantine Hoeger and Peterson provides background in chemistry and biochemistry with a relatable context to ensure students of all disciplines gain an appreciation of chemistry's significance in everyday life. Known for its clarity and concise presentation, this book balances chemical concepts with examples drawn from students' everyday lives and experiences to explain the quantitative aspects of chemistry and provide deeper insight into theoretical principles. The Seventh Edition focuses on making connections between General Organic and Biological Chemistry through a number of new and updated features, including all new Mastering Reactions boxes, Chemistry in Action boxes, new and revised chapter problems that strengthen the ties between major concepts in each chapter, practical applications, and much more. NOTE: this is just the standalone book. If you want the book access card, order the ISBN below: 032175011X 9780321750112 *Fundamentals of General Organic and Biological Chemistry Plus MasteringChemistry with eText Access Card Package*. Package consists of 0321750837 9780321750839 *Fundamentals of General Organic and Biological Chemistry* 0321776461 9780321776464 *MasteringChemistry with Pearson eText Valuepack Access Card for Fundamentals of General Organic and Biological Chemistry*.

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*The Electron* Robert Andrews Millikan, 1917

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*General Chemistry* Ralph H. Petrucci,F. Geoffrey Herring,Jeffrey D. Madura,Carey Bissonnette,2010-05

**Pearson Chemistry Queensland 11 Skills and Assessment Book** Elissa Huddart,2018-10-04 Introducing the Pearson Chemistry 11 Queensland Skills and Assessment Book Fully aligned to the new QCE 2019 Syllabus Write in Skills and Assessment Book written to support teaching and learning across all requirements of the new Syllabus providing practice application and consolidation of learning Opportunities to apply and practice performing calculations and using algorithms are integrated throughout worksheets practical activities and question sets All activities are mapped from the Student Book at the recommend point of engagement in the teaching program making integration of practice and rich learning activities a seamless inclusion Developed by highly experienced and expert author teams with lead Queensland specialists who have a working understand what teachers are looking for to support working with a new syllabus

**Chemistry in Context** AMERICAN CHEMICAL SOCIETY.,2024-04-11

**Solving General Chemistry Problems** Robert Nelson Smith,Willis Conway Pierce,1980-01-01

**Modern Analytical Chemistry** David Harvey,2000 This introductory text covers both traditional and contemporary topics relevant to analytical chemistry Its flexible approach allows instructors to choose their favourite topics of discussion from additional coverage of subjects such as sampling kinetic method and quality assurance

## **Calculating Average Atomic Mass Worksheet Answers** Book Review: Unveiling the Power of Words

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